Crystal growth and characterization of the cubic semiconductor Cu₂SnSe₄

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X-ray powder diffraction study of the *p*-type semiconductor Cu_2SnSe_4 shows that this material crystallizes in the cubic structure, space group $F\overline{4}3m$, with unit cell parameter a = 5.6846(3) Å. The temperature variation of the hole concentration between 120 and 300 K, obtained from the Hall effect and electrical resistivity measurements, is due to the thermal activation of an acceptor level with ionization energy of about 0.02 eV. The temperature variation of the hole mobility is explained by considering the scattering of charge carriers by ionized impurities and acoustic phonons. From this analysis, the density-of-states effective mass of the holes is estimated to be about 0.8 m_e , m_e being the free electron effective mass. From the optical absorption spectra, the fundamental absorption edge is found to be direct. The value of the lowest energy gap and the spin-orbit splitting were estimated to be about 0.35 and 0.20 eV, respectively. The temperature dependence of the magnetization measurements shows that Cu_2SnSe_4 is paramagnetic, indicating that most of the copper atoms have the divalent charge state. © 2002 American Institute of Physics. [DOI: 10.1063/1.1492018]

I. INTRODUCTION

The low-melting point $A_2^IB^{IV}C_3^{VI}$ ternary compounds 1 are of considerable interest $^{2-8}$ as potential small band-gap semiconductors for electro-optic devices.² One of them, Cu₂SnSe₃, which crystallizes in a monoclinic structure^{7,8} could be used for acousto-optic applications^{6,7} in the infrared region around 0.8 eV. However, as in the case of most of the $A_2^I B^{IV} C_3^{VI}$ semiconductors, the preparation of homogeneous crystals of this compound is not easy and deviation from the ideal composition is observed in samples prepared from different conventional crystal growth techniques. In an attempt to improve the quality of the synthesized crystals, ingots with slight deviations from the ideal stoichiometry were prepared from the melt. Se-rich crystals thus obtained, as analyzed by x-ray powder diffraction (XRD) and energy dispersive x-ray (EDX) spectroscopy, show a cubic crystalline structure and compositions of Cu:Sn:Se very close to 2:1:4, respectively. This indicates the formation of Cu₂SnSe₄ crystals. The compound Cu₂SnSe₄, or more precisely $Cu_2Sn\square Se_4$, where \square denotes the cation vacancy which is included to maintain the same number of cations and anions sites, belongs to one of the families of fourfold defect derivative of the A^{II}B^{VI} binary semiconductors.⁹ It is expected, in order to maintain an average of four valence electrons per lattice site, that copper atoms should be divalent (Cu^{2+}) in Cu_2SnSe_4 .

The crystal structure of Cu_2SnSe_4 was earlier reported by Bok and De Witt¹⁰ (Joint Committee on Powder Diffraction Standards No. 16-0670). They found that this material crystallizes in the cubic structure, space group $F\overline{4}3m$, with lattice parameter a = 5.690(2) Å. More recently,^{11,12} analysis of XRD patterns of nanocrystals of Cu₂SnSe₄, synthesized through solvothermal process, also indicates that the compound crystallizes in a cubic structure with lattice constant a = 5.712 Å. ¹¹ In this article we report on the growth, crystal structure, electrical, optical, and magnetic characterization of Cu₂SnSe₄. The study shows that this material is a *p*-type cubic semiconductor and presents a direct band gap.

II. EXPERIMENTAL DETAILS

Ingots of Cu₂SnSe₄ were prepared from the melt by the vertical Bridgman–Stockbarger technique. They were obtained by heating the mixture of Cu, Sn and Se, in the proportion 2:1:4 sealed in evacuated quartz ampoules. The ampoules were placed in a multiple zone vertical furnace. Initially they were heated from room temperature to 1150 °C at a rate of 40 °C/h. The molten mixture was kept at this temperature for 24 h. In order to assure a homogeneous mixing, the ampoule was agitated periodically. It was later cooled at a rate of 10 °C/h up to 800 °C, then at 1 °C/h to 640 °C. The cooling rate from 640 to 500 °C was 5 °C/h. The ingots were annealed at this temperature for 120 h. The furnace was then turned off and the ingot cooled down to room temperature.

The chemical analysis of samples taken from the central part of the ingots was performed by EDX using a Kevex model Delta-3 system connected to a Hitachi model S-2500

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scanning electron microscope (SEM). The error in standard-less analysis was around 5%.

X-ray powder diffraction (XRD) data were collected at room temperature in reflection mode using a Siemens D5005 diffractometer equipped with a Cu-target tube and a diffracted beam graphite monochromator. A fixed aperture and divergence slit of 1 mm, a 0.1 mm monochromator slit, and 0.6 mm detector slit were used. The specimen was scanned in the 2θ range of $20^{\circ}-120^{\circ}$, the scan step was 0.02° , and the time of counting in every step was 45 s. Quartz was used as an external standard. Precise determination of the peak positions was performed using conventional analytical software.

The hole concentration p and Hall mobility μ in the temperature range from 77 to 300 K were determined by combined Hall effect and conductivity measurements made by the four point probe standard technique. In-soldered contacts were made by etching the sample's surface with a low melting point solder flux using a low power iron solder. The contacts were ohmic in the temperature range of interest. A magnetic field of 10 KG was employed in the Hall measurements.

The absorption coefficient spectra at room temperature in the energy range from 0.1 to 1.0 eV was obtained using a Perkin-Elmer Fourier transform infrared (FTIR) spectrometer model 1725X.

The magnetic properties were studied in the temperature range from 5 to 300 K by magnetization measurements using a Quantum Design MPMS-5 superconducting quantum interference device magnetometer with an external field of 200 G.

III. RESULTS AND DISCUSSION

A. Chemical composition and x-ray diffraction analysis

As obtained from EDX, the typical chemical compositions of Cu:Sn:Se samples used for the electrical and optical studies were 28.6:14.5:56.9 atomic percentage, very close to the ideal value 2:1:4. However, there is a slight excess of Sn with respect to Cu (Cu/Sn \approx 1.97) and of cations over Se (metal/Se \approx 0.76). The x-ray powder pattern of Cu₂SnSe₄ is shown in Fig. 1. The indexing of the XRD data was performed by means of the computer program DICVOL91.¹³ The ten peak positions measured from the XRD pattern were used as input data. A unique solution was readily obtained in



FIG. 1. XRD powder pattern of Cu_2SnSe_4 . Values of (h,k,l) for each diffraction line observed are indicated.

a cubic system with unit cell parameter a = 5.6846(3) Å. The entire XRD pattern was reviewed by using the program *NBS*AID583*,¹⁴ in the cubic space group *F43m*. Table I shows the XRD data for Cu₂SnSe₄, which included the de Wolff, M_N ,¹⁵ and Smith–Snyder, F_N ,¹⁶ figures of merit. The Cu₂SnSe₄ structure can be described as derivative of the zinc blende where Zn atoms are replaced by Cu, and Sn atoms, and the cationic vacancy. The atomic positions are given in Table II.

B. Hole concentration and mobility

As checked by a thermal probe, all the samples obtained from the ingots were *p*-type conductivity. The variation of $\ln p$ of Cu₂SnSe₄ as a function of $10^3/T$ is shown in Fig. 2. It is observed that *p* decreases with increasing temperature up to about 120 K. Then, it increases with *T*. The appearance of a minimum in the temperature variation of the hole concentration was also observed in Cu₂SnSe₃ and is indicative of impurity band conduction in the low temperature region.⁷

The increases of p with T in the high temperature region (T>120 K) is probably due to the thermal activation of a

TABLE I. Observed and calculated x-ray powder diffraction data for Cu₂SnSe₄. The unit cell lattice parameter is a = 5.6846(3) Å. Space group F43m (T²_d, No. 216), Z = 1. The de Wolff, M_N , and Smith–Snyder, F_N , figures of merit are $M_{10} = 216.9$; and $F_{10} = 46.3$ (0.0015, 14), respectively.

$2\theta_{\rm obs}(^{\circ})$	$d_{\rm obs}({ m \AA})$	$(I/I_o)_{\rm obs}$	hkl	$2\theta_{\rm calc}(^{\circ})$	$d_{\text{calc}}(\text{\AA})$	$\Delta 2 \theta$ (°)
27.137	3.2831	100	111	27.147	3.2820	0.010
45.077	2.0095	34.8	220	45.070	2.0098	-0.007
53.415	1.7138	16.5	311	53.410	1.7140	-0.005
65.677	1.4204	4.9	400	65.639	1.4211	-0.038
72.418	1.3039	5.1	331	72.402	1.3041	-0.016
83.189	1.1603	4.9	422	83.181	1.1604	-0.008
89.499	1.0941	3.3	511	89.508	1.0940	0.009
100.098	1.0048	2.9	440	100.080	1.0049	-0.018
106.540	0.9611	3.6	531	106.569	0.9609	0.029
117.942	0.8989	3.3	620	117.954	0.8988	0.012

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TABLE II. Atomic coordinates of the Cu_2SnSe_4 compound in the F43m space group.

Atom	Site	x	У	z	foc
Cu	4a	0	0	0	0.50
Sn	4a	0	0	0	0.25
\square^{a}	4a	0	0	0	0.25
Se	4 <i>c</i>	$\frac{1}{4}$	$\frac{1}{4}$	$\frac{1}{4}$	1.0

^aCation vacancy.

shallow acceptor level. In this case, for nearly uncompensated conductivity, the hole concentration in the valence band p_v is given by the expression¹⁷

$$p_v(T) \approx A(T) \exp(-E_A/2K_B T), \tag{1}$$

where E_A is the thermal impurity-to-valence band activation energy and A a parameter which varies slowly with T.

To estimate E_A from the p(T) curve, the variation of p between 120 and 300 K was analyzed by using Eq. (1). A linear plot of $\ln p$ vs $10^3/T$, as shown in Fig. 2, was observed in this temperature range. From the slope of this line, the activation energy of the acceptor level is estimated to be $E_A \approx 20$ meV.

The variation of the hole mobility μ with temperature is shown in Fig. 3. It can be noted that at low temperatures μ slightly increases with increasing *T* indicating that the holes are predominantly scattered by ionized impurities. However, at high temperatures, μ decreases with *T* indicating that acoustic phonon scattering is the dominant process in this temperature range.

To estimate the strength of different scattering mechanisms in the valence band, μ can be calculated by using Mathiessen's rule

$$\mu^{-1} = \mu_I^{-1} + \mu_{AC}^{-1}, \tag{2}$$



FIG. 2. Plot of $\ln p$ of Cu_2SnSe_4 vs $10^3/T$ in the temperature range from 80 to 300 K. The temperature variation of holes in the valence band calculated from Eq. (1) is also shown by the continuous line. From the slope of this line, the activation energy of the acceptor level is estimated to be around 20 meV.



FIG. 3. Variation with temperature of the experimental hole mobility in the valence band of Cu₂SnSe₄. The theoretical fit to the $\mu(T)$ data, by using Eqs. (2)–(4) with the fixed parameters $\rho \approx 5.1 \text{ g/cm}^3$, $v \approx 3.5 \times 10^5 \text{ cm/s}$, $\epsilon_o \approx 24$, and $N_1 \approx 2 \times 10^{21} \text{ cm}^{-3}$, is shown by a continuous line. The parameters obtained from the fit were $m_{hd}^* \approx 0.8 m_e$ and $E_{ac} \approx 30 \text{ eV}$. Separate contributions to the total mobility due to ionized impurities and acoustic phonons are also shown by dotted and dashed lines, respectively.

where μ_I and μ_{AC} represent the mobilities of the charge carriers due to the scattering by ionized impurities and acoustic-lattice modes.

For *p*-type samples, the Hall mobility due to ionized impurity scattering is calculated according to the Brooks–Herring formula¹⁸

$$\mu_{I} = (2/300)2^{7/2} \epsilon_{o}^{2} (K_{B}T)^{3/2} [\pi^{3/2} e^{3} (m_{\rm vd}^{*})^{1/2} N_{1}f(x)]^{-1}.$$
(3)

Here ϵ_o is the static dielectric constant, m_{hd}^* the hole densityof-states effective mass, N_I the ionized impurity concentration, and the function f(x) is given by the relation $\ln(1+x)$ -x/(1+x), where $x = 6 \epsilon_o m_{hd}^* (K_B T)^2 / \pi e^2 h^2 p$.

The acoustical-mode scattering mobility is given by¹⁸

$$\mu_{AC} = (2/300)(8\pi)^{1/2} e\hbar^4 \rho v^2 [3E_{\rm ac}^2(m_{\rm hd}^*)^{5/2}(k_BT)^{3/2}]^{-1},$$
(4)

where $\rho \approx 5.1 \text{ g/cm}^3$ is the x-ray mass density, $v \approx 3.5 \times 10^5 \text{ cm/s}$ the longitudinal velocity of sound, and E_{ac} the valence band deformation potential.

The temperature dependence of μ was fitted to Eqs. (2)– (4). In the fit, $m_{\rm hd}^*$ and $E_{\rm ac}$ were considered as adjustable parameters. From the electrical data, the ionized impurity concentration $N_I \approx N_A \approx p$ was estimated to be around 2 $\times 10^{21}$ cm⁻³. Also, from the expression $\epsilon_o^2 \propto 1/E_o$ given by Li, Anderson, and Plovnick¹⁹ the dielectric constant was estimated to be $\epsilon_o \approx 24$, by using $E_o \approx 0.35$ eV for the energy gap, as determined from the optical data (to be discussed later). Hence, with the values of ρ , v, N_I , and ϵ_o given above, a good fit, as shown in Fig. 3 by continuous lines, is obtained with $m_{\rm hd}^* \approx 0.8m_e$ and $E_{\rm ac} \approx 30$ eV. These values are nearly of the same order of magnitude as $1.2 m_e$ and 21 eV, respectively, obtained⁷ for Cu₂SnSe₃.

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FIG. 4. Absorption coefficient spectra at room temperature of Cu₂SnSe₄ in the energy range from 0.1 to 0.9 eV. Optical transitions from the highest and split-off valence bands to the conduction band are indicated as E_o and E_o + Δ_o , respectively. The peak at around 0.18 eV is attributed to a valence band-to-acceptor level transition. The fit of the expression $\alpha \sim (hv - E_A)^{3/2}/(hv)^3$ to the data of α in the range from 0.10 to 0.26 eV, which gives $E_A \approx 85$ meV, is given in this figure by the continuous line.

C. Optical absorption coefficient

The optical absorption coefficient was obtained from the measured transmittance through the well-known relation $\alpha = (1/t)\ln(I/I_o) - \alpha_C$, where *t* is the thickness of the sample, I_o and *I* the intensity of the incident and transmitted light through the sample, respectively, and α_C is a term that accounts for the losses of the energy at the sample's surface.

Figure 4 shows the absorption coefficient spectra α of Cu_2SnSe_4 in the energy range from 0.1 to 0.9 eV at 300 K. Three absorption peaks with their maxima at about 0.18, 0.42, and 0.88 eV, are observed. In the cubic symmetry, due to the spin-orbit splitting, the threefold degenerate p-like Γ_{15} valence band separates into a twofold Γ_8 and single Γ_7 bands. It is thus expected that two optical transitions from the highest doubly degenerate Γ_8 and spin-orbit split-off Γ_7 valence bands to the conduction band (Γ_6) should occur at $\mathbf{k}=0$. Thus, the last two absorption bands can be associated with these two transitions, respectively. Also, since in a direct gap semiconductor it is expected that $\alpha hv \sim (hv)$ $-E_o)^{1/2}$ for photon energy $hv \ge E_o$, to verify the nature of the band gap of Cu₂SnSe₄, we plot $(\alpha hv)^2$ as a function of hv in Fig. 5. A linear dependence is observed for both the last two transitions indicating that Cu₂SnSe₄ has a direct allowed band gap at $\mathbf{k}=0$. The values $E_o \approx 0.35$ eV and E_o $+\Delta_o \approx 0.55 \text{ eV}$ at 300 K, were estimated by extrapolating the line portion of the curves to $(\alpha h v)^2 = 0$. This gives for the spin-orbit splitting the value $\Delta_{a} \approx 0.2$ eV. This is in good agreement with that calculated for Cu₂SnSe₃ and Cu₂GeSe₃ which is 0.3 and 0.24 eV, respectively.²

On the other hand, it was found that the narrow absorption band with the peak at about 0.18 eV follows a dependence of the form $\alpha \sim (hv - E_A)^{3/2}/(hv)^3$. This corresponds to a valence band-to-acceptor level transition²⁰ The fit of this expression to the data of α in the range from 0.10 to 0.26 eV, as also shown in Fig. 4 by the continuous line, gives E_A



FIG. 5. Plot of $(\alpha hv)^2$ vs hv of Cu₂SnSe_a at room temperature in the energy range from 0.2 to 0.9 eV. From the extrapolation of the line portion of the curves to $(\alpha hv)^2=0$, the values $E_o \approx 0.35$ eV and $E_o + \Delta_o \approx 0.55$ eV were estimated. This gives the value $\Delta_o \approx 0.2$ eV for the spinorbit splitting at $\mathbf{k}=0$.

 \approx 85 meV. This value is about five times higher than E_A \approx 20 meV estimated from electrical data. This suggests that at least two acceptor defect states are present in Cu₂SnSe₄. In can be also noted that in the hydrogenic approximation, the activation energy of a single shallow acceptor state can be estimated from the expression $E_{A0} \approx 13.6 \ (m_{\rm hd}^*/m_e)/\epsilon_o^2$. In the present case, using $m_{\rm hd}^*$ and ϵ_o given above, we obtain $E_{A0} \approx 20$ meV, which coincides with the value obtained from the electrical data. This suggests that this shallow acceptor level is due to the cooper vacancy acting as a single acceptor state (V_{Cu}^{1-}) . This identification is consistent with EDX analysis which indicates that our samples are Cu poor. On the other hand, for a double acceptor state,²¹ two levels at $E_{A1} \approx 1.7 \ E_{A0} \approx 35 \text{ meV}$ and $E_{A2} \approx 4 \ E_{A0} \approx 80 \text{ meV}$ are expected in this compound. The last value is in good agreement with $E_A \approx 85$ meV, obtained from the optical data, indicating that this level is probably due to a double acceptor state. The possible origin of this level will be discussed in the next section.

D. Magnetization measurements

The data of the temperature dependence of zero-fieldcooled (ZFC) and field-cooled (FC) magnetization M for an applied field H = 200 G for Cu₂SnSe₄ are shown in Fig. 6. As observed. M_{ZFC} curve slightly increases with increasing temperature, whereas M_{FC} increases with decreasing T. This constitutes a characteristic sign of irreversible magnetic behavior and indicates that a remarkable hysteresis appears in this compound below 300 K. We have no explanation for these results at the present time. However, it is observed that both M_{ZFC} and M_{FC} have positive values in all the ranges of temperature studied showing that Cu₂SnSe₄ is paramagnetic. Because Cu¹⁺ should take a diamagnetic $3d^{10}$ state, it is suggested that most of the Cu atoms are divalent (Cu²⁺), with $3d^9$ configuration, which should lead to a paramagnetic behavior. This is consistent with the fact that copper atoms



FIG. 6. Temperature dependence of the magnetization of Cu_2SnSe_4 , for an applied field H = 200 G, in the range from 5 to 300 K. Zero-field-cooled (ZFC) and field-cooled (FC) magnetization are indicated by open and closed circles, respectively.

should be divalent in Cu₂SnSe₄ in order to maintain an average of four valence electrons per lattice site. Also consistent with EDX analysis, the double acceptor level at about 80 meV can thus be explained as due to the presence of divalent copper vacancies (V_{Cu}^{2-}) . However, the possibility that a fraction of the Cu atoms are monovalent cannot be discarded because of the presence of the shallow single acceptor level at 20 meV attributed to copper vacancies (V_{Cu}^{1-}) . This also explains the small value of the observed susceptibility as due to the contribution of an appreciable fraction of Cu¹⁺ ions with a diamagnetic $3d^{10}$ state.

IV. CONCLUSIONS

In summary, XRD study of *p*-type Cu₂SnSe₄ shows that this material crystallizes in a cubic structure, space group $F\overline{4}3m$, with unit cell parameter a=5.6846(3) Å. From the temperature variation of hole concentration properties of this material it is found that above about 120 K, the carrier concentration is dominated by a shallow acceptor level with an activation energy of about 20 meV. The origin of this level, consistent with simple theoretical considerations for single acceptor defect states and chemical composition measurements, is tentatively assigned to single copper vacancies (V_{Cu}^{1-}) . From the analysis of the mobility data, by assuming the scattering by acoustic phonons and ionized impurities, the density of state hole effective mass and the valence band deformation potential are calculated to be around $0.8 m_e$ and 30 eV, respectively. On the other hand, from the optical absorption coefficient spectra at room temperature, the energy of the band gap and the spin-orbit splitting are estimated to be 0.35 and 0.2 eV, respectively. From the magnetization measurements it is found that Cu₂SnSe₄ is paramagnetic. This behavior is attributed to the fact that most of the copper atoms in this compound are expected to be divalent. The acceptor level of about 80 meV observed from optical data is thus attributed as due to the presence of divalent copper vacancies.

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